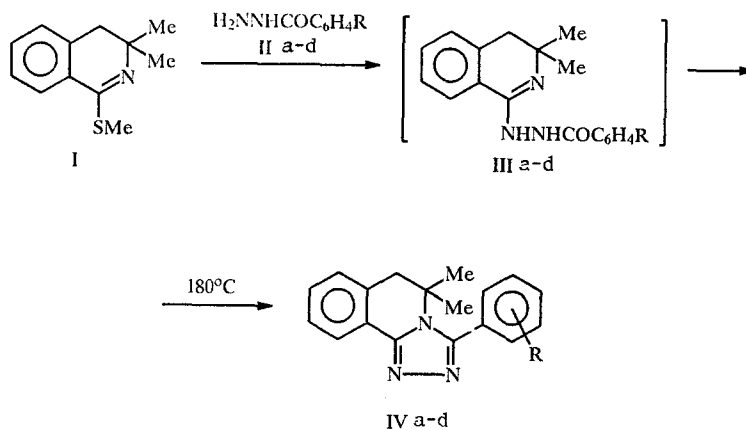


3-ARYL-5,5-DIMETHYL-5,6-DIHYDRO-1,2,4-TRIAZOLO[3,4-*a*]ISOQUINOLINES: A NEW EXAMPLE OF ATROPO ISOMERISM

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On refluxing in *o*-dichlorobenzene with equimolar amounts of benzoic acid hydrazides IIa-d the previously described thiolactim ether I [1] forms triazoloisoquinolines IVa-d. The reaction proceeds through a step involving the formation of [N'-(R-benzoyl)hydrazino]-3,3-dimethyl-3,4-dihydroisoquinolines IIIa-d; this was demonstrated thoroughly in the case of IIa \rightarrow IIIa \rightarrow IVa in the case of stepwise carrying out of the reaction in methanol (30 min, refluxing) and *o*-dichlorobenzene (3 h).

In the PMR spectra of IVa, b the signals of the protons of the 6-H proton and the protons of the 5-CH₃ groups are observed in the form of singlets, while for compounds with substituents R in the ortho position (IVc, d), the protons of the methyl groups give two singlet signals, whereas the CH₂ group of IVd gives a spectrum of the AB type.



II-IV ^aR=H, ^bR=4-Br, ^cR=2-NO₂, ^dR=2-Me

A study of IV by molecular mechanics (with the aid of the Alchemy II program) shows that the 3-aryl substituent cannot be located in the same plane (coplanar) as the plane of the triazoloisoquinoline system because of the presence of methyl groups attached to the C₍₅₎ atom, even if R = H, and free rotation about the C₍₃₎-C_(1') bond is impossible. In accordance with this, the unsymmetrically [relative to the C₍₃₎-C_(1') axis] substituted IV, including IVc, d, should be produced in the form of a racemic mixture of atropo isomers. Thus, the magnetic nonequivalence of the groups of 5-CH₃ and 6-H protons in the *o*-R-substituted triazoloisoquinolines IVc, d is explained by their diastereotopic character.

Compound IIIa (C₁₈H₁₉N₃O). This compound had mp 220-222°C (methanol). IR spectrum (mineral oil): 1557, 1575, 1609, 1645, 1655, 3170 cm⁻¹ (broad); 3270 cm⁻¹ (broad). PMR spectrum (d₆-DMSO): 1.21 (6H, s, 2CH₃), 2.83 (2H, s, CH₂), 6.7-8.2 (8H, m, aromatic H), 10.19 (1H, broad s, NH). The yield was 53%.

Compound IVa (C₁₈H₁₇N₃). This compound had mp 210-211.5°C (with sintering). IR spectrum (in mineral oil): 1595, 3060 cm⁻¹. PMR spectrum (d₆-DMSO): 1.25 (6H, s, 2CH₃), 7.41 (3H, m, 7-9-H), 7.54 (5H, s, 2'-6'-H), 8.03 (1H, m, 10-H). The yield was 84%.

Compound IVb (C₁₈H₁₆BrN₃). This compound had mp 292-295°C (from the reaction mixture). PMR spectrum (d₆-DMSO): 1.25 (6H, s, 2CH₃), 3.05 (2H, s, CH₂), 7.25-7.83 (7H, m, aromatic H), 8.03 (1H, m, 10-H). The yield was 94%.

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Compound IVc ($C_{18}H_{16}N_4O_2$). This compound had mp 194-195.5°C [2-propanol—methanol (2:1)]. PMR spectrum (d_6 -DMSO): 1.22 (3H, s, CH_3), 1.30 (3H, s, CH_3), 3.13 (2H, s, CH_2), 7.49 (3H, m, 7-9-H), 7.91 (3H, m, 4'-6'-H), 8.04 (1H, m, 10-H), 8.30 (1H, dd, $J_{3',4'} = 8.1$, $J_{3',5'} = 1.9$ Hz, 3'-H). The yield was 94%.

Compound IVd ($C_{19}H_{19}N_3$). This compound had mp 157-158°C [heptane—2-propanol—benzene (5:2:1)]. PMR spectrum (d_6 -DMSO): 1.10 (3H, s, 5- CH_3); 1.23 (3H, s, 5- CH_3); 2.13 (3H, s, 2'- CH_3); 3.02, 3.22 (2H, CH_2 , $J_{AB} = 16.4$ Hz); 7.39 (7H, m, aromatic H); 8.04 (1H, m, 10-H). The yield was 91%.

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